

A new protection/activation strategy for the synthesis of naturally occurring and non-natural α -N-alkylamino acids

Review Article

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Summary. A new method for the preparation of N-methylamino acids and some of their derivatives starting from hexafluoroacetone protected amino acids is described. The new concept results in saving of steps compared to conventional protection/activation techniques. Protection and deprotection proceed without racemization.

Keywords: Amino acids – Hexafluoroacetone – α -N-Methylamino acids – α -N-Phosphinoylmethylamino acids – Pro-Glu-chimeras – Pro-Tau-chimeras

Introduction

The development of new strategies for the synthesis of non-proteinogenic and non-natural amino acids is of current interest (O'Donnell, 1988; Williams, 1989; Duthaler, 1994). Some represent versatile building blocks for the construction of peptide mimetics, glycopeptide mimetics, depsipeptide mimetics, synthetic enzymes, new drugs and agrochemicals (Toniolo, 1990; Giannis and Kolter, 1993; Liskamp, 1994; Gante, 1994). Furthermore, they are valuable starting materials for combinatorial chemistry.

Recently, we have developed a new protection/activation concept for the derivatization of multifunctional amino acids (Burger et al., 1995; Windeisen et al., 1995; Pires et al., 1996). Hexafluoroacetone reacts with α -amino acids to give 2,2-bis(trifluoromethyl) substituted oxazolidin-5-ones in high yield. This heterocyclization process results in a simultaneous protection of the α -amino and α -carboxylic group. Functional groups present in the side chain, like -COOH and -OH remain unaffected. Consequently, this method can be applied for regioselective functional group manipulation of multifunctional α -amino acids (Burger and Rudolph, 1990; Pires et al., 1996).

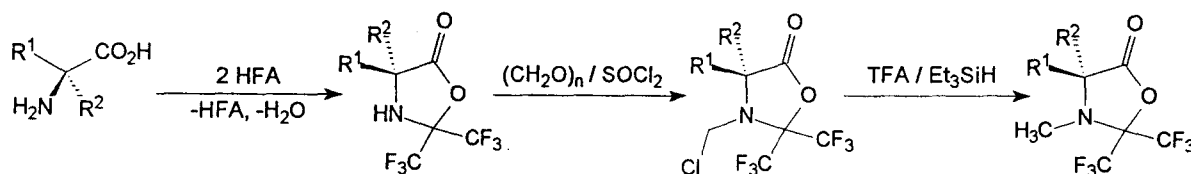
Deprotection of the α -carboxy and the α -amino group can be achieved simultaneously under mild conditions (water/2-propanol, room temperature). Likewise, aminolytic ring opening reactions are coupled with the deprotection of the amino group. Therefore, the new concept results in saving of steps compared to conventional protection/activation techniques. Furthermore, protection and deprotection proceed without racemization.

Finally, the presence of the fluoroalkyl substituents enables monitoring the progress of the protection, the functional group transformation and the deprotection process by ^{19}F NMR spectroscopy. We now disclose, that this new protection/activation strategy is also applicable for the synthesis of various N-substituted α -amino acids.

Synthesis of α -N-methylamino acids

α -N-Methylamino acids are constituents of various peptides and depsipeptides isolated from plant strains, microorganisms and marine species. Some of them exhibit highly interesting therapeutic profiles (Chruma et al., 1997). Incorporation of α -N-methylamino acids into strategical positions of peptides and depsipeptides leads to an enhanced proteolytic stability, to an increase in lipophilicity and to a stabilization of secondary structure domains. Furthermore, certain α -N-methylamino acids itself are biologically active compounds (Paruszewski et al., 1996; Okamoto and Quastel, 1997). Consequently, a number of synthetic routes to optically pure α -N-methylamino acids have been developed (Effenberger et al., 1983; Oppolzer et al., 1993; Dorow and Gingrich, 1995; Ebata et al., 1996; Bowman and Coghlan, 1997; Chruma et al., 1997; Muller et al., 1997; Bhatt et al., 1997).

In a three component reaction hexafluoroacetone protected α -amino acids, paraformaldehyde and thionyl chloride react to give α -N-chloromethylamino acid derivatives. The compounds obtained are readily transformed on treatment with triethylsilane/trifluoroacetic acid into α -N-methylamino acid derivatives. Both steps can be performed as one-pot procedure without solvent in nearly quantitative yield (Spengler and Burger, 1998a). When phosphorus tribromide is used instead of thionyl chloride, the corresponding N-bromomethyl compounds are available (Spengler and Burger, 1998b). The latter can be transformed into the iodo compounds applying the Finkelstein protocol.



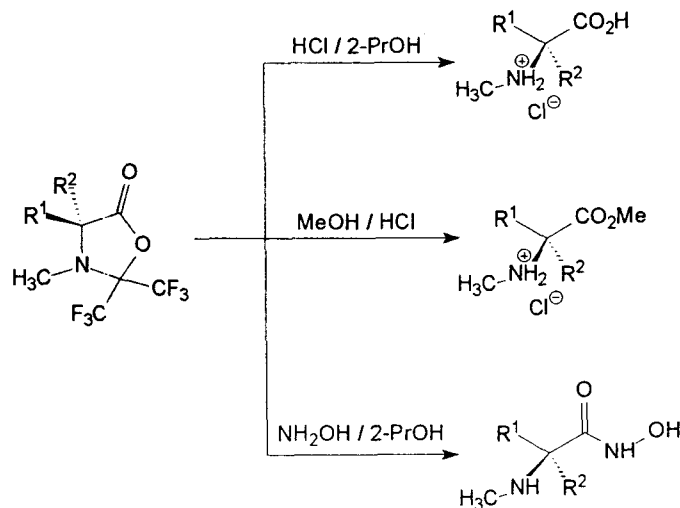
$\text{R}^1 = \text{H, alkyl, aryl}; \text{R}^2 = \text{H, aryl}$

Scheme 1

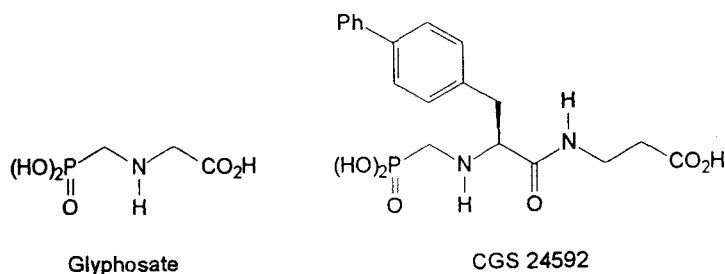
Since the N-methylated amino acid derivatives are carboxylic group activated species, they readily can be deprotected or derivatized to give the unprotected, optically active α -N-methylamino acids, ester hydrochlorides, amides, dipeptides and hydroxamic acids, respectively.

N-Phosphinoylmethylamino acids

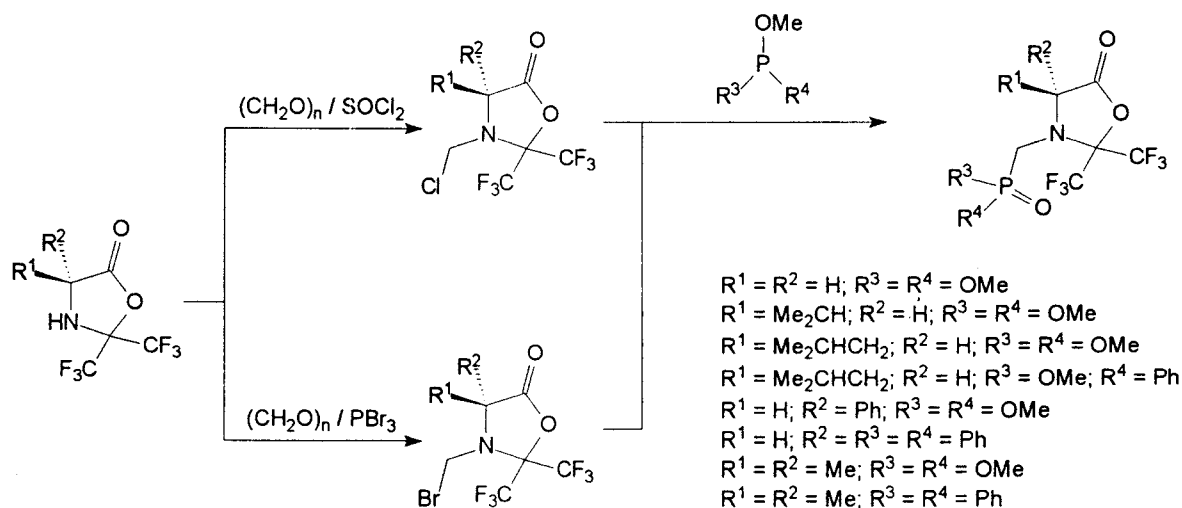
N-Phosphorylated peptides are naturally occurring species. Phosphoramidon isolated from *Actinomycetes* is a powerful inhibitor of the Zn metalloprotease thermolysin (Murano et al., 1982; Gómez-Monterrey et al., 1992). However, the biological half-life is short because of the phosphorus-nitrogen bond is sensitive to hydrolysis. A methylene spacer has been introduced between phosphorus and nitrogen to overcome this drawback. This special class of amino acids has potential as building block for the synthesis of new drugs, e.g. CGS 24592 (De Lombaert et al., 1994) and agrochemicals. N-Phosphonomethyl glycine ("glyphosate") inhibits the shikimic acid pathway in plants and



Scheme 2



Scheme 3



Scheme 4

represents the most widely used herbicide today (Sikorski and Gruys, 1997; Pfliegel et al., 1977).

With P(III)-species possessing at least one alkoxide ligand, N-halomethyl oxazolidin-5-ones undergo Michaelis-Arbusov reaction (Spengler and Burger, 1998b). While N-bromomethyl and N-iodomethyl compounds react exothermically within minutes, the N-chloromethyl compounds need several days at room temperature until the reactions are complete.

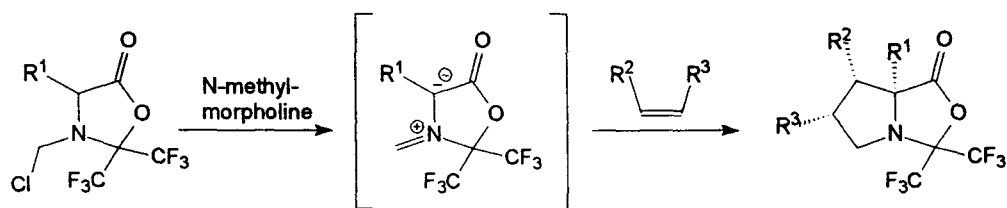
As carboxyl group activated species they can be readily deprotected or transformed into optically active amides, peptides, azapeptides and hydroxamic acids, respectively (Spengler and Burger, 1998b). The nucleophilic ring opening and the deblocking of the α -amino group are simultaneously occurring reactions. The over-all reaction sequence is suitable for generating libraries of N-phosphinoylmethylamino acid derivatives, including peptide- and depsipeptide mimetics.

New conformationally restricted phosphorus and sulfur containing structure mimetics of glutamic acid and taurine Pro-Glu-chimeras, Pro-Tau-chimeras

Conformationally restricted amino acids are important tools for the construction of biological active compounds with improved selectivity. In this context, proline analogues possessing the characteristics of other amino acids ("chimeras") are of special interest (Langlois and Andriamialisoa, 1991; Baldwin et al., 1995). Replacement of natural amino acids in biologically active peptides by proline-Xaa-chimeras has led to better understanding of their bioactive conformation (Plucinska et al., 1993). E.g. the neuroexcitatory activity of kainic acid is intrinsically associated to its action as conformationally constrained glutamate (Carpes et al., 1997). Since L-glutamate is a major

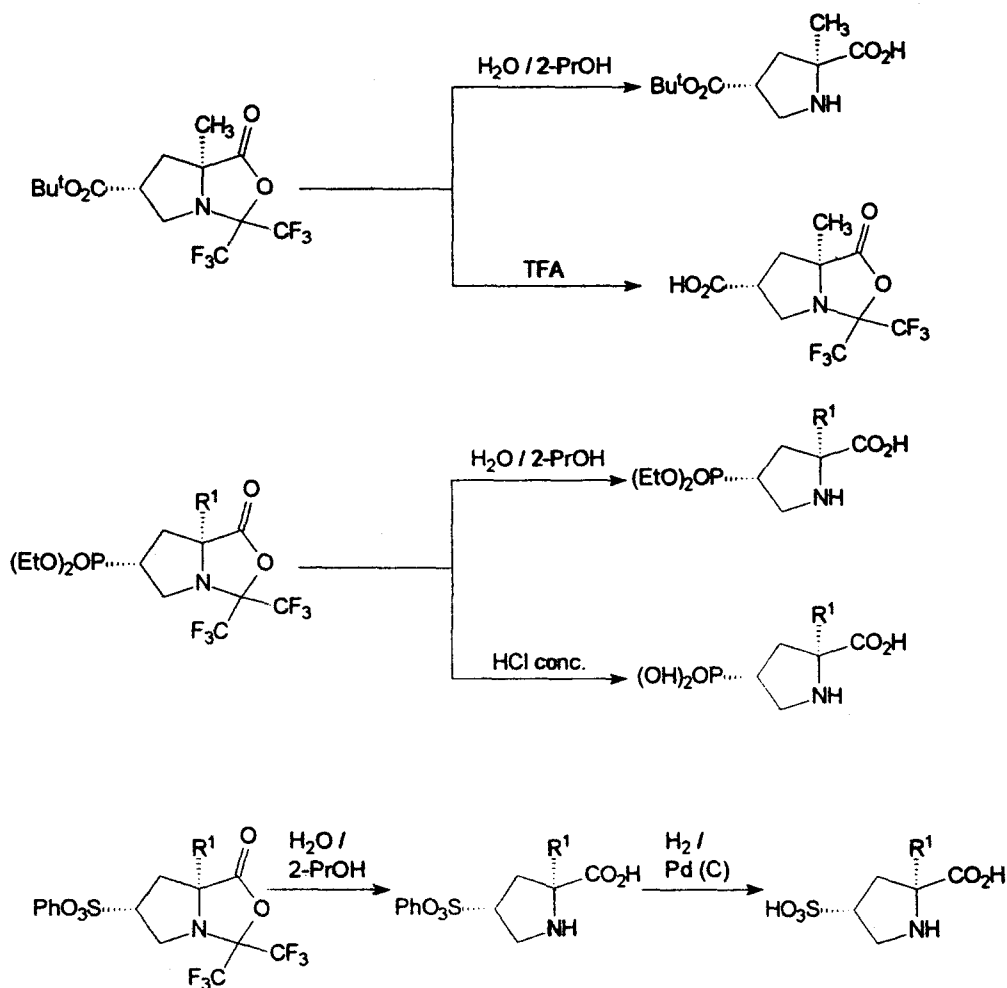


Among the many routes so far developed for the synthesis of proline-Xaa-chimeras, the concept of the 1,3-dipolar cycloaddition (Huisgen, 1984) turned out to be the most powerful one in respect of generating substituent patterns showing a maximum in variability (Gordon et al., 1996). Azomethine ylides are generated from N-halomethyl substituted 2,2-bis(trifluoromethyl)-



R¹ = H, alkyl, aryl; R² = H, CO₂Bu^t; R³ = H, PO(OEt)₂, SO₂OC₆H₅, CO₂Bu^t

Scheme 6



Scheme 7

oxazolidin-5-ones on treatment with *tert.* amines like N-methylmorpholine. The 1,3-dipoles obtained via 1,3-elimination can be trapped *in situ* by a large number of dipolarophiles providing highly functionalized proline derivatives.

The [3+2] cycloaddition proceeds via a concave transition state, the two five-membered ring systems being *cis* fixed. Therefore, the relative stereochemistry can be controlled efficiently. The products obtained are racemates. The progress of the [3+2] cycloaddition process can be monitored by ^{19}F NMR spectroscopy.

With *tert.* butyl acrylate two regioisomers are formed, which can be separated by flash chromatography. The cycloadducts represent Pro-Glu and Pro-Asp chimeras, respectively. In contrast, diethyl vinylphosphonate and phenyl vinylsulfonate on reaction with the 1,3-dipoles provide the corresponding cycloadducts regio- and stereospecifically, which represent conformationally rigid glutamic acid analogues (Schedel et al., 1999). Furthermore, the sulfur-containing proline derivative mimics a conformationally rigid taurine. Multifunctional proline derivatives are structurally interesting scaffolds for drug design and potential secondary structure mimetics.

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